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(54) **PROCESS FOR THE PRODUCTION OF BIODEGRADABLE HIGH PERFORMANCE
HYDROCARBON BASE OILS**

**VERFAHREN ZUR HERSTELLUNG VON BIOLOGISCH ABBAUBARE
HOCHLEISTUNGS-KOHLLENWASSERSTOFF-BASISÖLE**

**PROCEDE DE PRODUCTION D'HUILES DE BASE HYDROCARBONEES BIODEGRADABLES ET
EXTREMEMENT EFFICACES**

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(56) References cited:
EP-A- 0 225 053 **EP-A- 0 321 307**
EP-A- 0 323 092 **US-A- 3 365 390**

EP 0 876 446 B1

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Description**1. Field of the Invention**

5 **[0001]** This invention relates to biodegradable high performance hydrocarbon base oils, suitable as engine oil and industrial oil compositions. In particular, it relates to lubricant base oil compositions, and process for making such compositions by the hydroisomerization/hydrocracking of paraffinic waxes, suitably Fischer-Tropsch waxes.

2. Background

10 **[0002]** It is well known that very large amounts of lubricating oils, e.g., engine oils, transmission oils, gear box oils, etc., find their way into the natural environment, accidentally and even deliberately. These oils are capable of causing much environmental harm unless they are acceptably biodegradable. For this reason there is increasing emphasis in this country, and abroad, to develop and employ high performance lubricant base oils which are environmentally friendly, or substantially biodegradable on escape or release into the environment.

15 **[0003]** Few hydrocarbon base oils are environmentally friendly though their qualities as lubricants may be unchallenged. The literature stresses the superior biodegradability of ester based lubricants, natural and synthetic, over hydrocarbon based products. However there is little or no emphasis on performance. Few references are found relating to the biodegradability of hydrocarbon lubricants. Ethyl Petroleum Additives's EP 468 109A however does disclose the biodegradability of lubricating oils containing at least 10 volume percent of a "biodegradable liquid hydrocarbon of lubricating viscosity formed by oligomerization of a 1-alkene hydrocarbon having 6 to 20 carbon atoms in the molecule and hydrogenation of the resultant oligomer." Apparently hydrogenated oligomers of this type have unexpectedly high biodegradability, particularly those having at least 50 volume percent dimer, trimer and/or tetramer. Ethyl Petroleum Additive's EP 558 835 A1 discloses lubricating oils having similar polyalphaolefin, PAO, components. However, both references point out performance debits for the synthetic and natural ester oils, such as low oxidative stability at high temperatures and poor hydrolytic stability. British Petroleum's FR 2675812 discloses the production of biodegradable PAO hydrocarbons base oils by dewaxing a hydrocracked base oil at low temperatures.

20 **[0004]** US-A-3365390 claims and discloses a process for producing lube oil which comprises: hydrocracking a deasphalted petroleum residuum boiling mostly above 800°F (426.7°C) and at least partially above 900°F (482.2°C) by contacting said residuum and hydrogen with a sulfactive hydrocracking catalyst in a hydrocracking zone under conditions to convert at least 20 percent of said residuum to distillates boiling lower than the feed and at least 30 percent of the portion of the said residuum boiling above 900°F (482.2°C) to distillates boiling below 900°F (482.2°C) and with a hydrogen consumption of at least 500 s.c.f. per barrel of residuum; separating the oil effluent from said hydrocracking zone into fractions including a distillate fuel and a hydrocracked lube oil boiling range fraction; dewaxing a said hydrocracked lube oil boiling range fraction, thereby obtaining a hydrocracked wax fraction of low nitrogen content; hydroisomerizing at least a portion of said hydrocracked wax fraction by contacting said wax fraction with an active reforming catalyst containing 0-2 weight percent-halide in a hydroisomerization zone under conditions to convert at least 20 percent of said wax fraction to distillates boiling below 750°F (398.9°C); and separating the oil effluent from said hydroisomerization zone into fractions including a distillate fuel and a hydroisomerized lube oil boiling range fraction.

40 **[0005]** EP-A-0323092 claims and discloses a process for producing a lubricating oil having a high viscosity index and a low pour point from a Fischer-Tropsch wax, which process comprises:

45 (a) contacting the Fischer-Tropsch wax with a hydrotreating catalyst (which may be unsulfided) and hydrogen in a hydrotreating zone (R-1) to reduce the oxygenate and trace metal levels of the wax and to partially hydrocrack and isomerize the wax;

50 (b) contacting the hydrotreated Fischer-Tropsch wax from step (a) with hydrogen in a hydroisomerization zone (R-2) in the presence of a fluorided Group VIII metal-on-alumina catalyst having (i) a bulk fluoride concentration ranging from about 2 to about 10 weight percent, wherein the fluoride concentration is less than about 3.0 weight percent at the outer surface layer to a depth less than one one-hundredth of an inch (0.254 mm), provided the surface fluoride concentration is less than the bulk fluoride concentration, (ii) an aluminum fluoride hydroxide hydrate level greater than 60 (e.g. at least about 100) wherein an aluminum fluoride hydroxide hydrate level of 100 corresponds to the X-ray diffraction peak height of 5.66 Å (0.566 nm) for a Reference Standard, and (iii) a N/Al ratio less than about 0.005;

55 (c) fractionating the effluent from step (b) in a fractional zone (F-1) to produce a lubricating oil fraction boiling above about 640°F (337.8°C) (e.g. above about 700°F (371.1°C)) at atmospheric pressure; and

(d) dewaxing the lubricating oil fraction from step (c) in a dewaxing zone (D-1) to produce a dewaxed lubricating oil having a viscosity index of at least 130 (e.g., at least 140) and a pour point less than about 0°F (-17.8°C) e.g. below -6°F (-21.1°C).

5 [0006] EP-A-0225053 claims and discloses a process for producing a lubricating oil stock with a target pour point and a high viscosity index by catalytically dewaxing a lube base stock containing waxy, paraffinic components with a dewaxing catalyst comprising at least one large pore zeolite having a silica:alumina ratio of at least 10:1 and a hydrogenation-dehydrogenation component, in the presence of hydrogen under conventional dewaxing conditions of temperature and pressure, to isomerize the waxy paraffinic components to relatively less waxy iso-paraffinic components, characterized by partial removal of waxy components to produce an intermediate product having a pour point at least 6°C above the target pour point, and selectively dewaxing the intermediate product by preferential removal of straight chain, waxy paraffinic components over iso-paraffinic components, to produce a lube oil stock product with the target pour point and having a high viscosity index.

10 [0007] EP-A-0321307 claims and discloses a process for producing lube oil base stocks or blending stocks having a pour point of about -21°C or lower and a viscosity index of about 130 and higher by the isomerization of wax, said process comprising (1) isomerizing the wax in an isomerization unit over an isomerization catalyst to a level of conversion such that about 40% or less unconverted wax, calculated as (unconverted wax)/(unconverted wax + dewaxed oil) x100 remains in the fraction of the isomerate boiling in the lube boiling range sent to the dewaxing unit, fractionating the total product from the isomerization zone into a lube fraction boiling in the lube boiling range and solvent dewaxing said fraction and (2) recovering a lube oil product having a VI of at least 130 and a pour point of -21°C or lower.

20 [0008] There is a clear need for biodegradable high performance hydrocarbon base oils useful as engine oil and industrial oil, or lubricant compositions which are at least equivalent to the polyalphaolefins in quality, but have the distinct advantage of being more biodegradable.

25 3. Summary of the invention

[0009] The invention, which supplies these and other needs, accordingly relates to a process for the production of a biodegradable high performance hydrocarbon base oil by hydrocracking and hydroisomerization of paraffinic, or waxy hydrocarbon feeds obtained from Fischer-Tropsch processes, all or at least a portion of which boils above 371°C (700°F).

30 [0010] The invention further provides the use of the biodegradable high performance hydrocarbon base oil obtained by such process as an engine oil or as an engine oil component or the use of such material to produce a medicinal grade white oil.

[0011] According to the process of the invention, the waxy feed is first contacted, with hydrogen, over a dual functional catalyst to produce hydroisomerization and hydrocracking reaction sufficient to convert from 20 to 50 %, preferably from 25 to 40 %, on a once through basis based on the weight of the 371°C+ (700°F+) feed, or 371°C+ (700°F+) feed component, to 371°C- (700°F-) materials, and produce 371°C+ (700°F+) material rich in isoparaffins.

[0012] The resultant crude product, which contains both 700°F- (371 °C-) and 700°F+(371°C) materials, characterized generally as a C₅-1050°F+ (566°C+) crude fraction, is first topped via atmospheric distillation to produce a lower boiling fraction the upper end of which boils between 650°F (343.3°C) and 750°F (398.9°C), e.g., 700°F (371°C), and a higher boiling, or bottoms fraction having an initial boiling point ranging between 650°F (343.3°C) and 750°F (398.9°C), e.g., 700°F (371°C), and an upper end or final boiling point of 1050°F+ (566°C+), e.g., a 700°F+ (371 °C+) fraction. The lower boiling fraction, e.g., the 700°F- (371 °C-) fraction, from the distillation is a non-lube, or fuel fraction.

40 [0013] At these conversion levels, the hydroisomerization/hydrocracking reactions convert a significant amount of the waxy, or paraffinic feed to 700°F+ (371°C+) methyl-paraffins, i.e., isoparaffins containing one or more methyl groups in the molecule, with minimal formation of branches of carbon number greater than 1; i.e., ethyl, propyl, butyl or the like. The 700°F+ (371 °C+) bottoms fractions so-treated contain 700°F+ (371°C+) isoparaffins having from 6.0 to 7.5 methyl branches per 100 carbon atoms, preferably from 6.5 to 7.0 methyl branches per 100 carbon atoms, in the molecule. These isoparaffins, contained in a mixture with other materials, provide a product from which high performance, highly biodegradable lube oils can be obtained.

50 [0014] The higher boiling bottoms fractions, e.g., the 700°F+(371°C+) bottoms fraction containing the methyl-paraffins, or crude fraction, is dewaxed in a conventional solvent dewaxing step to remove n-paraffins, and the recovered dewaxed product, or dewaxed oil, is fractionated under vacuum to produce paraffinic lubricating oil fractions of different viscosity grades, including hydrocarbon oil fractions suitable as high performance engine oils and engine lubricants which, unlike most hydrocarbon base oils, are biodegradable on release or escape into the environment. In terms of their performance they are unsurpassed by the PAO lubricants, and are superior thereto in terms of their biodegradability.

4. Detailed Description

[0015] The feed materials that are isomerized to produce the lube base stocks and lubricants with the catalyst of this invention are waxy feeds, i.e., C_5^+ , preferably having an initial boiling point above 350°F (176°C), more preferably above 550°F (288°C), and contain a major amount of components boiling above 700°F (371 °C) obtained from a Fischer-Tropsch process which produces substantially normal paraffins.

[0016] Fischer-Tropsch waxes are feed materials having negligible amounts of aromatics, sulfur and nitrogen compounds. The Fischer-Tropsch liquid, or wax, is characterized as the product of a Fischer-Tropsch process wherein a synthetic gas, or mixture of hydrogen and carbon monoxide, is processed at elevated temperature over a supported catalyst comprised of a Group VIII metal, or metals, of the Periodic Table of The Elements (Sargent-Welch Scientific Company, Copyright 1968), e.g., cobalt, ruthenium, iron, etc. The Fischer-Tropsch wax contains C_5^+ , preferably C_{10}^+ , more preferably C_{20}^+ paraffins. A distillation showing the fractional make up (± 10 wt.% for each fraction) of a typical Fischer-Tropsch process liquid feedstock is as follows:

Boiling Temperature Range	Wt. % of Fraction
IBP - 320°F (160°C)	13
320 - 500°F (160-260°C)	23
500 - 700°F (260-371°C)	19
700 - 1050°F (371-566°C)	34
1050°F+ (566°C+)	11
	<u>100</u>

[0017] The wax feed is contacted, with hydrogen, at hydrocracking/hydroisomerization conditions over a bifunctional catalyst, or catalyst containing a metal, or metals, hydrogenation component and an acidic oxide support component active in producing both hydrocracking and hydroisomerization reactions. Preferably, a fixed bed of the catalyst is contacted with the feed at conditions which convert 20 to 50 wt. %, preferably 25 to 40 wt. %, of the 700°F (371°C) components of the feed to 700°F- (371°C-) materials and produce a lower boiling fraction having an upper end boiling point between 650°F (343.3°C) and 750°F, e.g., 700°F (371°C), and a higher boiling, or bottoms fraction having an initial boiling point between 650°F (343.3°C) and 750°F (389.9°C), e.g., 700°F, the higher boiling fraction that remains containing high quality blending components for the production of high performance biodegradable base oils. In general, the hydrocracking/ hydroisomerization reaction is conducted by contacting the waxy feed over the catalyst at a controlled combination of conditions which produce these levels of conversion; i.e., by selection of temperatures ranging from 400°F (204°C) to 850°F (454°C), preferably from 500°F (260°C) to 700°F (371°C), pressures ranging generally from 100 pounds per square inch gauge (psig) to 1500 psig, preferably from 300 psig (21.1 Kg/cm²) to 1000 psig (70.31 Kg/cm²), hydrogen treat gas rates ranging from 1000 SCFB (178 m³/m³) to 10,000 SCFB (1780 m³/m³), preferably from 2000 SCFB (356 m³/m³) to 5000 SCFB (890 m³/m³), and space velocities ranging generally from 0.5 LHSV to about 10 LHSV, preferably from 0.5 LHSV to 2.0 LHSV.

[0018] The active metal component of the catalyst is a non-noble Group VIII metal, or metals, of the Periodic Table Of The Elements (Sargent-Welch Scientific Company Copyright 1968) in amount sufficient to be catalytically active for hydrocracking and hydroisomerization of the waxy feed. The catalyst may also contain, in addition to the Group VIII metal, or metals, a Group IB and/or a Group VIB metal, or metals, of the Periodic Table. Generally, metal concentrations range from 0.1 percent to 20 percent, based on the total weight of the catalyst (wt.%), preferably from 0.1 wt. percent to 10 wt. percent. The group VIII used in the invention are non-noble Group VIII metals as nickel and cobalt, or mixtures of these metals with each other or with other metals, such as copper, a Group IB metal, or molybdenum, a Group VIB metal. The metal, or metals, is incorporated with the support component of the catalyst by known methods, e.g., by impregnation of the support with a solution of a suitable salt or acid of the metal, or metals, drying and calcination.

[0019] The catalyst support is constituted of metal oxide, or metal oxides, components at least one component of which is an acidic oxide active in producing olefin cracking and hydroisomerization reactions. The catalyst support used in the present invention is constituted of silica and alumina, the content of silica being up to 35 wt.% The supports preferably constituted of from 2 wt.% to 35 wt.% silica, and has the following pore-structural characteristics:

Pore Radius (Å) 10 ⁻¹⁰ m	Pore Volume
0-300	>0.03 ml/g
100-75,000	<0.35 ml/g
0-30	<25% of the volume of the pores with 0-300(Å) 10 ⁻¹⁰ m radius

(continued)

Pore Radius (Å) 10 ⁻¹⁰ m	Pore Volume
100-300	<40% of the volume of the pores with 0-300(Å) 10 ⁻¹⁰ m radius

The base silica and alumina materials can be, e.g., soluble silica containing compounds such as alkali metal silicates (preferably where $\text{Na}_2\text{O}:\text{SiO}_2 = 1:2$ to $1:4$), tetraalkoxy silane, orthosilic acid ester, etc.; sulfates, nitrates, or chlorides of aluminum alkali metal aluminates; or inorganic or organic salts of alkoxides or the like. When precipitating the hydrates of silica or alumina from a solution of such starting materials, a suitable acid or base is added and the pH is set within a range of about 6.0 to 11.0. Precipitation and aging are carried out, with heating, by adding an acid or base under reflux to prevent evaporation of the treating liquid and change of pH. The remainder of the support producing process is the same as those commonly employed, including filtering, drying and calcination of the support material. The support may also contain small amounts, e.g., 1-30 wt.%, of materials such as magnesia, titania, zirconia or hafnia.

[0020] Support materials and their preparation are described more fully in U.S. Patent No. 3,843,509 incorporated herein by reference. The support materials generally have a surface area ranging from 180-400 m²/g, preferably 230-375 m²/g, a pore volume generally of 0.3 to 1.0 ml/g, preferably 0.5 to 0.95 ml/g, bulk density of generally 0.5-1.0 g/ml, and a side crushing strength of about 0.8 to 3.5 kg/mm.

[0021] The hydrocracking/hydroisomerization reaction is conducted in one or a plurality of reactors connected in series, generally from 1 to 5 reactors; but preferably the reaction is conducted in a single reactor. The waxy hydrocarbon feed, Fischer-Tropsch wax, preferably one boiling above 700°F (371°C), or has a large amount of 700°F+ (371°C) hydrocarbon components, is fed, with hydrogen, into the reactor, a first reactor of the series, to contact a fixed bed of the catalyst at hydrocracking/hydroisomerization reaction conditions to hydrocrack, hydroisomerize and convert at least a portion of the waxy feed to products which include after further work up high quality oils and lube blending components.

[0022] The following examples are illustrative of the more salient features of the invention. All parts, and percentages, are given in terms of weight unless otherwise specified.

Examples 1-9

[0023] A mixture of hydrogen and carbon monoxide synthesis gas ($\text{H}_2:\text{CO}$ 2.11-2.16) was converted to heavy paraffins in a slurry Fischer-Tropsch reactor. A titania supported cobalt rhenium catalyst was utilized for the Fischer-Tropsch reaction. The reaction was conducted at 422-428°F (217-220°C), 287-289 psig (20.18-20.32 Kg/cm²), and the feed was introduced at a linear velocity of 12 to 17.5 cm/sec. The alpha of the Fischer-Tropsch synthesis step was 0.92. The paraffinic Fischer-Tropsch product was isolated in three nominally different boiling streams; separated by utilizing a rough flash. The three boiling fractions which were obtained were: 1) a C₅-500°F (260°C) boiling fraction, i.e., F-T cold separator liquids; 2) a 500-700°F (260-371 °C) boiling fraction, i.e., F-T hot separator liquids; and 3) a 700°F+ (371 °C+) boiling fraction, i.e., a F-T reactor wax.

[0024] A series of base oils were prepared in runs made by hydrocracking and isomerizing the 700°F+ (371°C+) Fischer-Tropsch reactor wax feedstock, with hydrogen, at different levels of conversion over a silica exsiccated cobalt-moly-nickel catalyst (CoO, 3.6 wt. %; MoO₃, 16.4 wt.%; NiO, 0.66 wt. %; on a SiO₂-Al₂O₃ support, 13.7 wt. % of which is silica); having a surface area of 270 m²/g, and pore volume < 30 mm equal to 0.43). A combination of reaction conditions, i.e., as relates to temperature, space velocity, pressure and hydrogen treat rate, was used to convert 30 wt. %, 35 wt. %, 45 wt. %, 50 wt. %, 58 wt. %, 67 wt. %, and 80 wt. % respectively, of the feedstock to materials boiling below 700°F (371°C), i.e., 700°F-(371 °C-). The conditions for each of the respective runs and the yields which were obtained for each are given in Table 1. The Table also lists the amounts of IBP-650°F (343.3°C) and 650°F+ (343.3°C+) products obtained by 15/5 distillation.

TABLE I
CONVERSION TO 371.1°C - (700°F-) wt. %

Operating Conditions	30	35	45	50	58	67	80
Temperature, (°F) °C	361 (681.9)	365 (689)	374 (705.2)	372 (701.5)	376.5 (709.7)	375 (707.1)	377.4 (711.4)
Space Velocity, LHSV	0.42	0.50	0.50	0.45	0.50	0.43	0.44
Pressure, (psig) bar.g	--	--	68.97 (1000)	--	--	--	--
H ₂ Treat Rate, (SCF/B)	--	--	444.7 (2500)	--	--	--	--
Yields (wt. % recovery) m ³ H ₂ /m ³							
C1-C4	1.17	0.73	1.73	2.11	2.14	2.43	3.70
C5-320°F (160°C)	5.48	3.11	9.68	9.75	9.48	14.93	23.10
320-550°F (160-260°C)	10.43	10.11	17.82	17.92	22.87	25.20	27.04
550-700°F (260-371°C)	20.48	23.94	21.88	24.63	27.81	28.01	30.21
700°F+ (371°C+)	62.44	62.11	48.89	45.59	37.70	29.43	15.93
15/5 Composite							
Distillation (wt. %)							
IBP-650°F (343.3°C)	32.25	26.71	37.46	44.26	48.35	59.80	67.77
650°F+ (343.3°C+)	67.75	73.29	62.54	55.74	51.65	40.20	32.23

[0025] A 343°C+ (650°F+) bottom fraction was recovered from the products obtained from each of the runs by atmospheric distillation, and then again fractionated under high vacuum to produce several viscosity grades of lubricant, viz. 60N, 100N, 175N and about 350-400N. The residual products were then subjected to solvent dewaxing to remove waxy hydrocarbons and lower the pour point to about -18°C (-32°F).

[0026] For each viscosity grade, the dewaxing conditions were held constant so that the effect of conversion level on dewaxing could be evaluated. The dewaxing conditions for 100N and 175N viscosity grades at the 30%, 50%, 67% and 80% conversion levels are given in Table 2.

Table 2

Dewaxing Conditions ¹		
	Viscosity Grade	
	100N	175N
30% Conversion		
Solvent:Oil Ratio	3:1	3:1
Filter Temp, °C	-21	-21
Pour Pt, °C	-18	-18
50% Conversion		
Solvent:Oil Ratio	3:1	3:1
Filter Temp, °C	-21	-21
Pour Pt, °C	-21	-21
67% Conversion		
Solvent:Oil Ratio	3:1	3:1
Filter Temp, °C	-21	-21
Pour Pt, °C	-15	-18
80% Conversion		
Solvent:Oil Ratio	3:1	3:1
Filter Temp, °C	-21	-21
Pour Pt, °C	-24	-24

¹ All dewaxings employed 100% methylisobutylketone, MIBK.

[0027] The physical properties, yields of dewaxed oil, DWO, and corresponding dry wax contents (both as wt. % on waxy feed) for each dewaxing in terms of the 100N and 175N viscosity grades at specific levels of conversion are given in Table 3.

TABLE 3
DEWAXED BASE OIL PHYSICAL PROPERTIES
VISCOSITY GRADES

	30% Conversion		50% Conversion		67% Conversion		80% Conversion	
	100N	175N	100N	175N	100N	175N	100N	175N
Dewaxed Oil Yield/Dry Wax	80.7/17.6	75.3/21.4	93.0/6.6	91.1/7.7	97/2.4	92/5.2	98/2.0	96.3/1.7
Content (wt. % on waxy feed)								
Pour/Cloud Pt., °C	-18/-14	-18/-14	-21/-14	-21/-17	-15/-7	-18/-14	-24/-21	-24/-21
Density @ 15°C, kg/dm	0.8143	0.8218	0.8153	0.8229	0.8147	0.8231	0.8160	0.8234
Refractive Index @ 20°C								
Viscosity, cSt								
@ 40°C	15.59	26.96	16.28	29.14	15.90	28.76	16.71	18.94
@ 100°C	3.81	5.59	3.86	5.77	3.77	5.68	3.85	5.61
Viscosity Index	141	153	133	145	129	143	124	136
GCD, °C								
IBP	346	380	343	390	347	394	351	393
5%	369	408	367	418	369	419	370	416
50%	426	471	424	473	421	469	421	466
95%	486	535	488	531	479	524	478	523
FBP	522	567	528	565	515	558	513	559

[0028] Nuclear magnetic resonance (NMR) branching densities for 100N base oils produced at 30%, 50%, 67%, and 80% levels, respectively, are given in Table 4. It will be observed that the lower levels of methyl branching occurs at the lower conversion levels; with the biodegradability of the oil increasing at the lower levels of conversion. Compositions of highest biodegradability are thus produced at the 30 wt.% level of conversion, and the next highest biodegradability compositions are produced at the 50 wt.% conversion level.

Table 4

100N Base Oil, ¹³ CNMR Branching Densities				
	----% Conversion----			
Base Oil	30	50	67	80
V.I.	141	133	129	124
Per 100 Carbons Methyl Groups (CH ₃ ⁻)	6.8	7.5	7.5	7.8

[0029] It is also found that the viscosity index, VI, decreases with increasing level of conversion for each specific viscosity grade. This is because base oils prepared at higher conversion levels tend to be more highly branched and consequently have lower viscosity indexes. For the 100N base oils, the VI ranges from 141 to 118. For the 175N oils, the corresponding VI range is 153 to 136, respectively. The 175N base oils have VIs which are also comparable to the commercial ETHYLFLO 166 which has a VI of 143. The VI of the 100N viscosity grade is comparable to the commercial ETHYLFLO 164 which has a VI of 125. For purposes of comparison, certain physical properties of the commercial 100N ETHYLFLO 164 and 175N ETHYLFLO 166 are presented in Table 5.

Table 5

ETHYLFLO™ 164 (Lot 200-128)	
Viscosity at 100°C, cSt	3.88 (3.88 X 10 ⁻⁶ m ² /sec)
Viscosity at 40°C, cSt	16.9 (16.9 X 10 ⁻⁶ m ² /sec)
Viscosity at -40°C, cSt	2450 (2450 X 10 ⁻⁶ m ² /sec)
Viscosity Index	125
Pour Point, °C	-70
Flash Point (D-92), °C	217
NOACK volatility, %	11.7
CEC-L-33-T-82	30%
ETHYLFLO™ 166 (Lot 200-122)	
Viscosity at 100°C, cSt	5.98 5.98 X 10 ⁻⁶ m ² /sec)
Viscosity at 40°C, cSt	30.9 30.9 X 10 ⁻⁶ m ² /sec)
Viscosity at -40°C, cSt	7830 (7830 X 10 ⁻⁶ m ² /sec)
Pour Point, °C	-64
Flash Point (D-92), °C	235
NOACK VOLATILITY, %	6.1
Viscosity Index	143
CEC-L-33-T-82	29%

[0030] To determine the biodegradability of the DWO base stocks, and lubricant compositions, tests were conducted in accordance with CEC-L-33-T-82, a test method developed by the Coordinating European Council (CEC) and reported in "Biodegradability Of Two-Stroke Cycle Outboard Engine Oils In Water: Tentative Test Method" pp 1-8 and incorporated herein by reference. The test measures the decrease in the amount of a substrate due to microbial action. It has been shown, as measured by CEC-L-33-T-82 that the DWO base stocks, and lubricant compositions produced in accordance with this invention are of biodegradability above about 50%, and 10 are generally above about 50% to about 90%, and higher, biodegradable.

Examples 10-13

[0031] The CEC-L-33-T-82 test was run to observe the biodegradation of the following samples over a 21 day period, to wit:

Samples:

A: Base Oil 100N, 30 wt.% Conv. - 1.5133 g/100 mL FREON
 B: Base Oil 100N, 50 wt.% Conv. - 1.4314 g/100 mL FREON
 C: Base Oil 100N, 67 wt.% Conv. - 1.5090 g/100 mL FREON
 D: Base Oil 100N, 80 wt.% Conv. - 1.5388 g/100 mL FREON
 X: VISTONE A30 - 1.4991 g/100 mL FREON
 (Positive Calibration Material)

[0032] Each of the tests were conducted using a FREON solvent, and the stock solutions used were standard as required by the test procedure.

[0033] The inoculum used was non-filtered primary effluent from the Pike Brook Treatment Plant in Bellemead, New Jersey. The inoculum was determined to have between 1×10^4 and 1×10^5 colony forming units/mL (CFU/mL) by Easicult-TCC dip slides.

[0034] Triplicate test systems for all test materials and Vistone A30 were prepared and analyzed on day zero for parent material concentration. All extractions were performed as described in the test procedure. The analyses were performed on the Nicolet Model 205 FT-IR. Triplicate test systems for samples B through X, in addition to poisoned systems of each sample were placed on orbital shakers and continuously agitated at 150 rpm in total darkness at $25 \pm 0^\circ\text{C}$ until day twenty-one. On day twenty-one the samples were analyzed for residual parent material. Sample "A" was also evaluated at the day seven interval to determine removal rate along with the above mentioned samples. Triplicate systems for "A" were prepared, extracted and analyzed after seven, fourteen and twenty-one days of incubation.

RESULTS

[0035]

100N BASE OILS		
SAMPLE Level of Conversion	% BIODEGRADATION (21 DAYS)	STANDARD DEVIATION, SD
A: Base Oil 30 wt.%	84.62	1.12
B: Base Oil 50 wt.%	77.95	0.86
C: Base Oil 67 wt.%	73.46	1.01
D: Base Oil 80 wt.%	73.18	2.34
E. ETHYLFLO 164	30.00	0.54
X: VISTONE A30	98.62	1.09
¹ Based on analysis of triplicate inoculated test systems and triplicate poisoned test systems.		

RATE STUDY SAMPLE A		
DAY	% BIODEGRADATION	SD
7	76.15	2.74
14	82.82	2.37
21	84.62	1.12

Examples 14-16

[0036] The CEC-L-33-T-82 test was run to observe the biodegradation of the following test materials over a 21 day period.

Samples:

A:¹ Base Oil 175N, 30 wt.% Conv. - 1.58 g/ 100 mL FREON

B:² Base Oil 175N, 50 wt.% Conv. - 1.09 g/100 mL FREON

C:¹ Base Oil 175N, 80 wt.% Conv. - 1.43 g/100 mL FREON

X:¹ VISTONE A30 - 1.5 g/100 mL FREON

(Positive Calibration Material)

[0037] Each of the tests were conducted using a FREON solvent, and the stock solutions used were standard as required by the test procedure.

[0038] The inoculum was non-filtered primary effluent from the Pike Brook Treatment Plant in Bellemead, New Jersey. The inoculum was determined to have between 1×10^4 and 1×10^5 colony forming units/mL (CFU/mL) by Easicult-TCC dip slides.

[0039] Triplicate test systems for all test materials and Vistone A30 were prepared and analyzed on day zero for parent material concentration. All extractions were performed as described in the test procedure. The analyses were performed on the Nicolet Model 205 FT-IR. Triplicate test systems for samples A through X, in addition to poisoned systems of each sample were placed inside environmental chambers and continuously agitated at 150 rpm in total darkness at $25 \pm 0^\circ\text{C}$ until day twenty-one. On day twenty-one the samples were analyzed for residual parent material.

RESULTS

[0040]

175N BASE OILS		
SAMPLE	% BIODEGRADATION (21 DAYS) ¹	SD
A: Base Oil	76.93	1.452
B: Base Oil	62.01	1.379
C: Base Oil	51.04	1.657
G. ETHYLFLO 166	29.0	
X: VISTONE A30	85.31	0.408

¹ Based on analysis of triplicate inoculated test systems and triplicate poisoned test systems.

[0041] These data show that two different 100N oils were of biodegradability approaching 75%, and two different 100N oils were of biodegradability well above 75%; one approximating 85%. The Blue Angels in Germany, defines "readily biodegradable" as >80% in the CEC-L-33-T-82 test. The three 175N oils that were demonstrated had biodegradability values ranging between about 51 % to about 77%.

[0042] The DWO base stocks, and lubricant compositions due to their high paraffinic content, > 97.5 Vol. %, are also suitable as feedstocks for medicinal grade white oils. The following is exemplary.

Example 18

[0043] A dewaxed 60N base oil was subjected to mild hydrofining over a Ni-Mn-MoSO₄ bulk catalyst to produce an 80 wt.% level of conversion (i.e., 240°C, 600 psi (42.4 Kg/cm²) H₂, 0.25 LHSV). The product readily passed the diagnostic "hot acid test" for medicinal grade white oils.

¹ 500 µL used to dose test systems to achieve ~7.5 mg loading of test material.

² 750 µL used to dose test systems to achieve ~7.5 mg loading of test material.

Claims

1. A process for the production of a biodegradable high performance hydrocarbon base oil which comprises

5 contacting on a once through basis a 371 °C+ (700°F+) paraffinic feed, or paraffinic feed containing 371 °C+ (700°F+) components, obtained from a Fischer Tropsch process, with hydrogen, over a dual functional catalyst, active for both hydrocracking and hydroisomerization, comprised of a non noble Group VIII metal, or metals, on a support constituted of silica and alumina the content of silica being up to 35 wt% and said support containing optionally from 1 to 30 wt % of magnesia, titania, zirconia or hafnia, to convert from 20 to 50 % based
10 on the weight of the 371°C+ (700°F+) feed, or 371°C+ (700°F+) feed components, to 371°C- (700°F-) material and to produce a crude fraction containing 371°C+ (700°F+) isoparaffins having from 6.0 to 7.5 methyl branches per 100 carbon atoms,

topping said crude fraction by atmospheric distillation to produce a residual bottoms fraction having an initial
15 boiling point in the range of from 343 to 399°C (650 to 750°F),

dewaxing said bottoms fraction with a solvent to recover a dewaxed oil, and

fractionating said dewaxed oil under a vacuum to recover said biodegradable high performance hydrocarbon
20 base oil.

2. The process of claim 1 wherein the catalyst is comprised of a Group IB or VIB metal, or metals, or both a Group IB and VIB metal, or metals, in-addition to the Group VIII metal, or metals.
- 25 3. The process of claim 2 wherein the concentration of the metal, or metals, ranges from 0.1 percent to 20 percent, based on the total weight of the catalyst, the Group IB metal is copper, the Group VIB metal is molybdenum, and the Group VIII metal is nickel or cobalt.
- 30 4. The process of claim 1 wherein the produced fraction containing 371°C+ isoparaffins having from 6.5 to 7.0 methyl branches per 100 carbon atoms in the molecules.
5. The process of claim 1 wherein the level of conversion of the 371 °C+ feed ranges from 25 to 40 % by weight.
- 35 6. Use of the biodegradable high performance hydrocarbon base oil obtained according to the process of claim 1 as engine oil or as engine oil component.
7. Use of the biodegradable high performance hydrocarbon base oil obtained according to the process of claim 1 to produce a medicinal grade white oil.

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Patentansprüche

1. Verfahren zur Herstellung von biologisch abbaubarem Hochleistungsöl auf Kohlenwasserstoffbasis, bei dem ein paraffinisches 371°C+ (700°F+)-Einsatzmaterial oder paraffinisches Einsatzmaterial, das 371°C+ (700°F+)
45 -Komponenten enthält, erhalten aus einem Fischer-Tropsch-Verfahren, auf Basis eines einmaligen Durchsatzes mit Wasserstoff über einem doppel funktionalen Katalysator kontaktiert wird, der sowohl für hydrierendes Cracken als auch für Hydroisomerisierung aktiv ist, aus einem NichtEdelmetall oder -metallen der Gruppe VIII auf einem Träger zusammengesetzt ist, der aus Siliciumdioxid und Aluminiumoxid zusammengesetzt ist, wobei der Siliciumdioxidgehalt bis zu 35 Gew.% beträgt und der Träger gegebenenfalls 1 bis 30 Gew.% Magnesiumoxid, Titandioxid,
50 Zirkoniumdioxid oder Hafniumoxid enthält, um 20 bis 50 %, bezogen auf das Gewicht des 371°C+ (700°F+)-Einsatzmaterials oder der 371°C+ (700°F+)-Einsatzmaterialkomponenten, in 371°C- (700°F-)-Isoparaffine mit 6,0 bis 7,5 Methylverzweigungen auf 100 Kohlenstoffatome umzuwandeln, die Rohfraktion durch atmosphärische Destillation getoppt wird, um eine Restsumpfproduktfraktion mit einem Anfangssiedepunkt im Bereich von 343 bis 399°C (650 bis 750°F) zu produzieren,
55 die Sumpfproduktfraktion mit Lösungsmittel entparaffiniert wird, um entparaffiniertes Öl zu gewinnen, und das entparaffinierte Öl unter Vakuum fraktioniert wird, um das biologisch abbaubare Hochleistungsöl auf Kohlenwasserstoffbasis zu gewinnen.

2. Verfahren nach Anspruch 1, bei dem der Katalysator aus Metall oder Metallen der Gruppe IB oder VIB, oder Metall oder Metallen sowohl der Gruppe IB als auch der Gruppe VIB, zusätzlich zu dem Metall oder den Metallen der Gruppe VIII zusammengesetzt ist.
- 5 3. Verfahren nach Anspruch 2, bei dem die Konzentration des Metalls oder der Metalle im Bereich von 0,1 % bis 20 % liegt, bezogen auf das Gesamtgewicht des Katalysators, das Metall der Gruppe IB Kupfer ist, das Metall der Gruppe VIB Molybdän ist, und das Metall der Gruppe VIII Nikkel oder Kobalt ist.
- 10 4. Verfahren nach Anspruch 1, bei der die produzierte Fraktion 371°C+-Isoparaffine mit 6,5 bis 7,0 Methylverzweigungen auf 100 Kohlenstoffatome in den Molekülen enthält.
5. Verfahren nach Anspruch 1, bei dem das Umwandlungsniveau des 371°C+ Einsatzmaterials im Bereich von 25 bis 40 Gew.% liegt.
- 15 6. Verwendung des biologisch abbaubaren Hochleistungsöls auf Kohlenwasserstoffbasis, das nach dem Verfahren gemäß Anspruch 1 erhalten wurde, als Motoröl oder als Motorölkomponente.
7. Verwendung des biologisch abbaubaren Hochleistungsöls auf Kohlenwasserstoffbasis, das nach dem Verfahren gemäß Anspruch 1 erhalten wurde war, zur Herstellung von Weißöl mit medizinischer Qualität.
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Revendications

- 25 1. Procédé pour la production d'une huile de base hydrocarbonée biodégradable de performances élevées, comprenant :

la mise en contact unique d'une charge paraffinique de 371 °C+ (700 °F+) ou d'une charge paraffinique contenant des composants de 371 °C+ (700 °F+), obtenue à partir d'un procédé de Fischer Tropsch, avec de l'hydrogène sur un catalyseur à double fonction, actif à la fois pour l'hydrocraquage et l'hydroisomérisation, constitué d'un ou plusieurs métaux non nobles du groupe VIII, sur un support constitué de silice et d'alumine, la teneur en silice allant jusqu'à 35 % en poids et ledit support contenant éventuellement 1 à 30 % en poids de magnésie, d'oxyde de titane, d'oxyde de zirconium ou d'oxyde d'hafnium pour convertir 20 à 50 %, par rapport au poids de la charge de 371 °C+ (700 °F+), ou des composants de charge de 371 °C+ (700 °F+), en un matériau de 371 °C- (700 °F-) et pour produire une fraction brute contenant des isoparaffines de 371 °C+ (700 °F+) ayant 6,0 à 7,5 ramifications méthyle pour 100 atomes de carbone, l'ététagage de ladite fraction brute par distillation atmosphérique pour produire une fraction résiduelle de bas de colonne ayant un point d'ébullition initial dans la plage de 343 à 399 °C (650 à 750 °F), le déparaffinage de ladite fraction de bas de colonne avec un solvant pour récupérer une huile déparaffinée, et le fractionnement de ladite huile déparaffinée sous vide pour récupérer ladite huile de base hydrocarbonée biodégradable de performances élevées.
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2. Procédé selon la revendication 1, dans lequel le catalyseur est composé d'un ou plusieurs métaux du groupe IB ou VIB ou d'un ou plusieurs métaux à la fois des groupes IB et VIB en plus d'un ou plusieurs métaux du groupe VIII.
- 45 3. Procédé selon la revendication 2, dans lequel la concentration en métal ou en métaux se situe dans la plage de 0,1 pour-cent à 20 pour-cent par rapport au poids total du catalyseur, le métal du groupe IB étant le cuivre, le métal du groupe VIB étant le molybdène et le métal du groupe VIII étant le nickel ou le cobalt.
- 50 4. Procédé selon la revendication 1, dans lequel la fraction produite contient des isoparaffines de 371 °C+ ayant 6,5 à 7,0 ramifications méthyle pour 100 atomes de carbone dans les molécules.
5. Procédé selon la revendication 1, dans lequel le taux de conversion de la charge de 371 °C+ se situe dans la plage de 25 à 40% en poids.
- 55 6. Utilisation de l'huile de base hydrocarbonée biodégradable obtenue selon le procédé de la revendication 1 comme huile de moteur ou comme composant d'huile de moteur.
7. Utilisation de l'huile de base hydrocarbonée obtenue selon le procédé de la revendication 1 pour la production

d'huile blanche de grade médical.

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